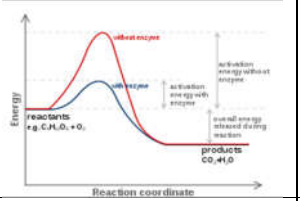
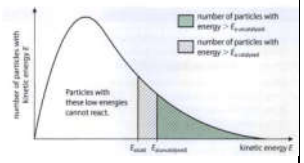
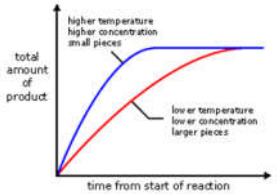
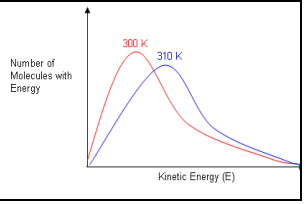
	1	2	3	4	5	6
1		Effective collisions	Activation energy	Maxwell-Boltzmann distribution	Rate of reaction	Q
2	How temperature affects rate	Q	Continuous monitoring method	Molecular distributions	Boil an egg or fry an egg?	Catalyst
3	Mass of reactant used up per second	How concentration affects rate		Q	Alternative reaction route	Collision frequency
4	$CaCO_3$ powder or lumps with HCl	Initial rate method	How surface area affects rate	Collision force	Q	How gas pressure affects rate
5	Q	Why most collisions do not result in a reaction	Storing milk in the fridge	Why a small increase in T has a large affect on the rate		Mass of product made per s
6	Rate and catalyst	Hydrochloric acid and sodium thiosulfate	Q	Molecular distributions	Why your book can burn, but is not currently on fire	

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